

Sequestration of Atmospheric Carbon Dioxide as Inorganic Carbon under Semi-Arid Forests

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Introductions



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Greenhouse Gases

Atmosphere contains ~320 billion tons CO₂



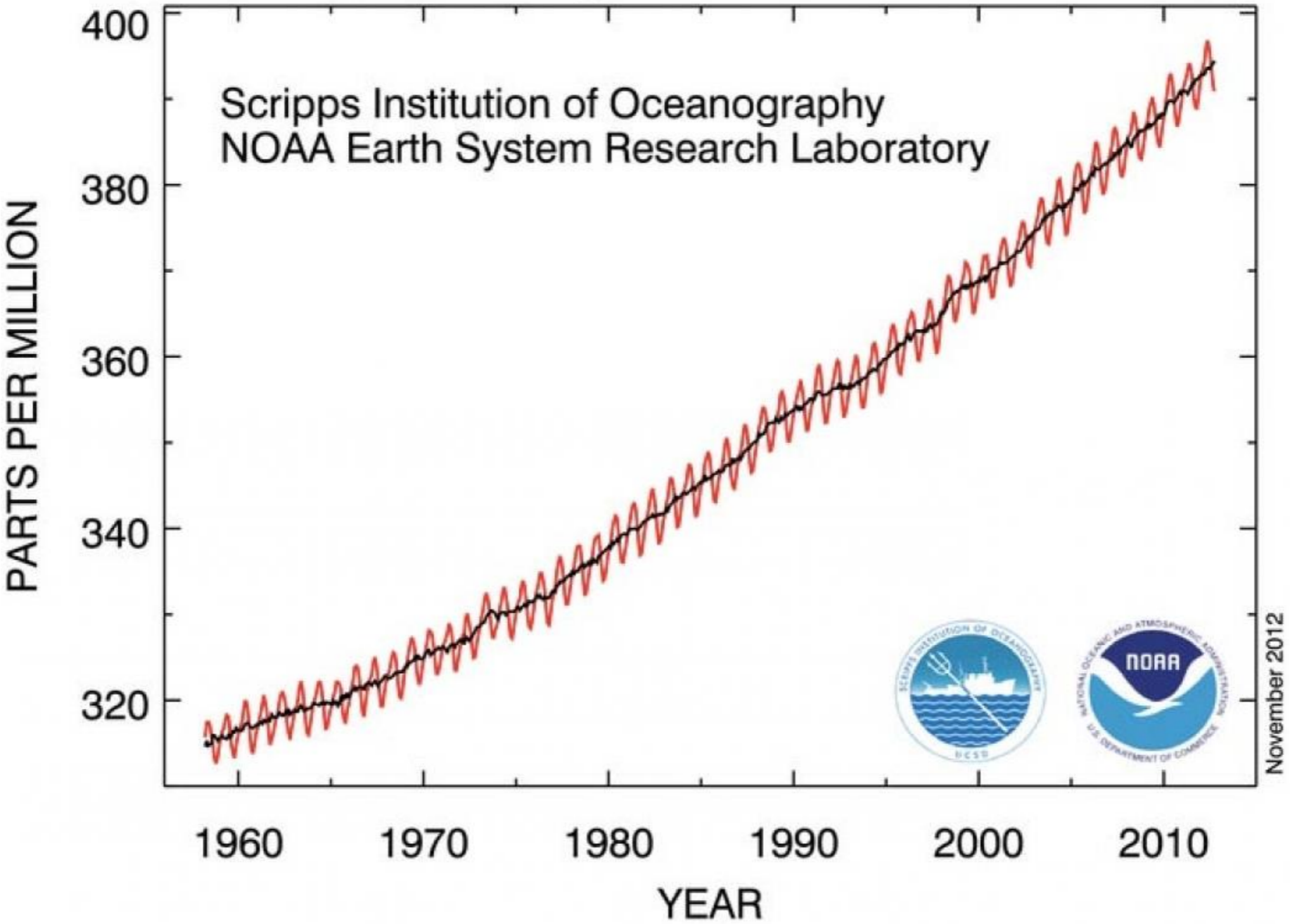
Annual increase is ~20 billion tons CO₂



Excessive CO₂ greenhouse gas may cause global warming, climate change, rising sea levels, ocean acidification, ocean deoxygenation, melting icefields, more forest fires, reduced food supply, etc.



Atmospheric CO₂ at Mauna Loa Observatory





What can be done?

- Burn Less Fossil Fuels. (difficult to achieve)
- Nuclear power, develop safer reactors, deal more effectively with nuclear wastes. (“Greens” opposition)
- Stop Deforestation
- Climate engineering projects proposed for *carbon dioxide removal (CDR)* and *solar radiation management (SRM)*. (large, expensive, controversial)
CDR captures CO₂ produced by large industrial plants, compresses it for transportation, and then injects it deep into a rock formation for permanent storage.
SRM projects seek to reduce global warming by reflecting sunlight, for example using stratospheric sulfate aerosols
- ***Carbon reduction & storage via forestation. (low cost & low tech)***

How to prevent temperature increase?



Pinatubo eruption June 1991



10 billion tons magma & 20 million tons SO_3



Global Temperatures decreased 0.5°C 1991-1993



Best not to count on volcanic eruptions!

How to prevent temperature increase?



Fossil fuels provide $\sim 80\%$ of global energy supply

Collect CO_2 and securely bury deep underground



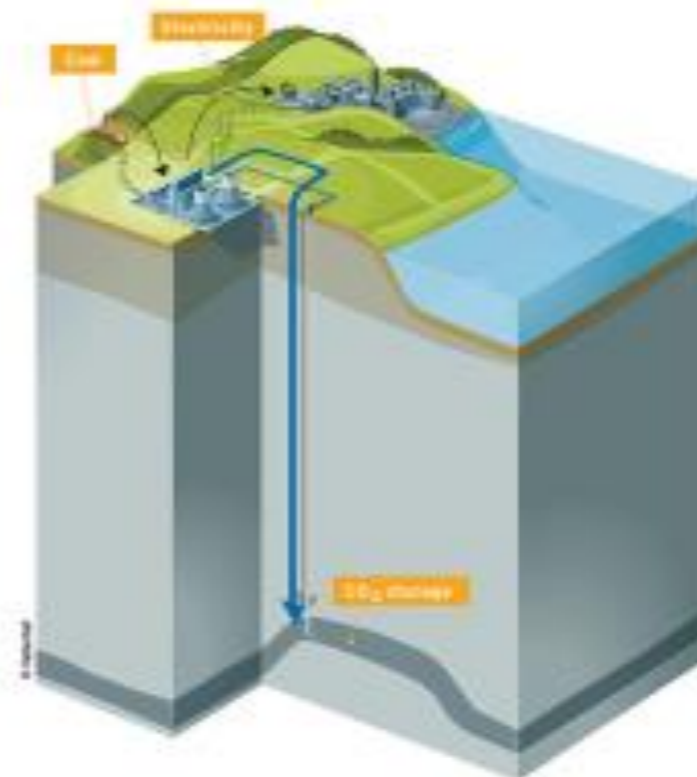
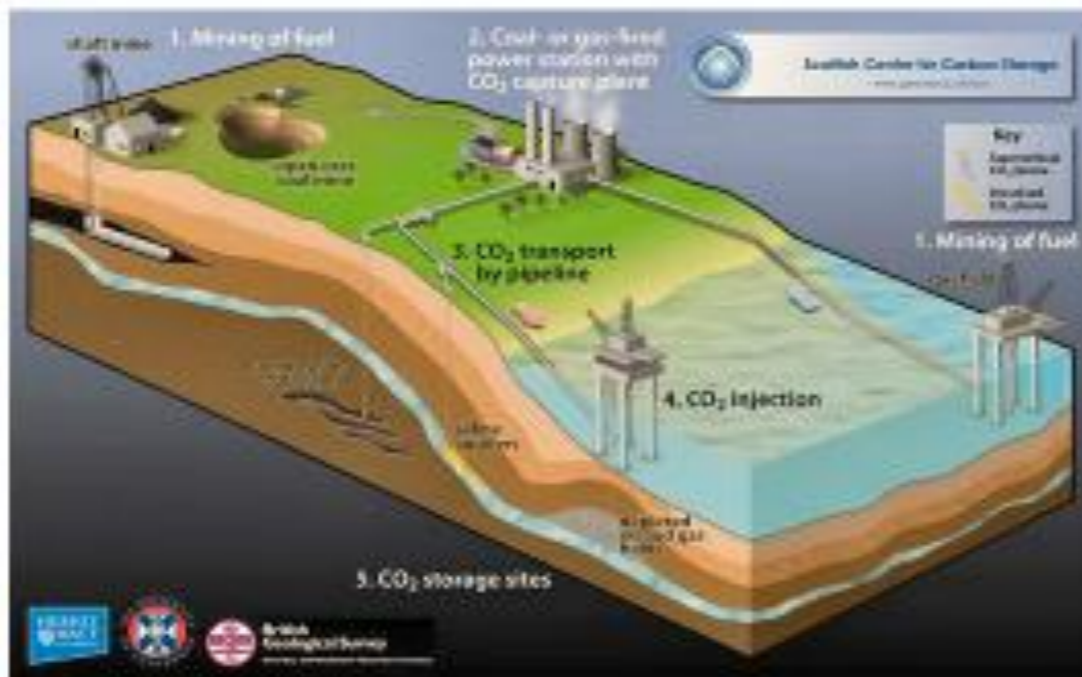
Sequester $\sim 40\%$ of annual CO_2 production



complicated & expensive

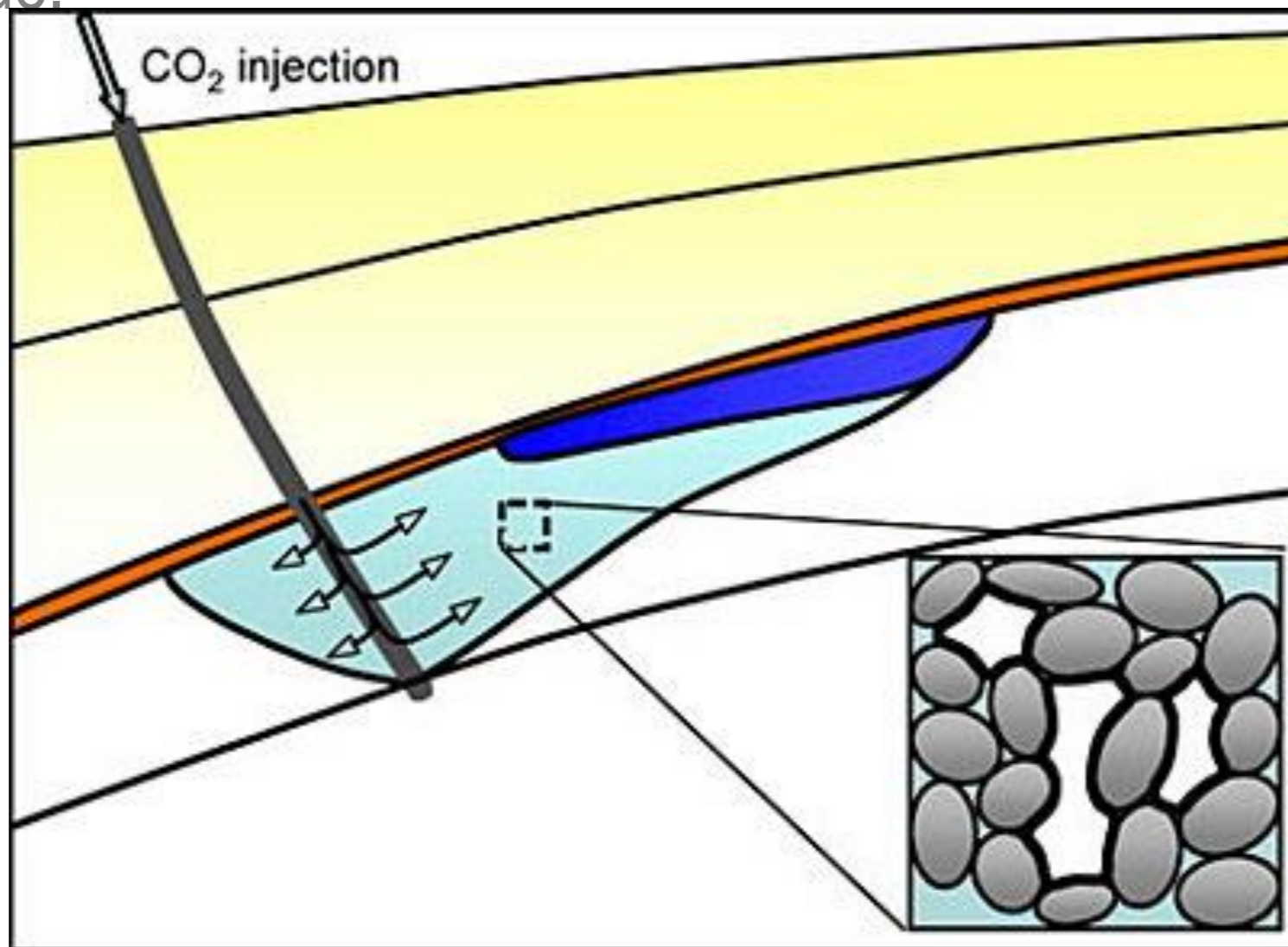
Carbon Capture and Storage: What It Is

- Capture CO₂ released from burning fuels and store it elsewhere
- Reduce CO₂ emission into atmosphere
- Steps:
 - 1.) Capture
 - 2.) Transport
 - 3.) Storage



Technological Feasibility

- Removing the Carbon Dioxide is not the problem, it's storing it that's the main issue.



How to prevent temperature increase?



develop and use safer nuclear power,
Molten Salt Breeder Reactors, etc.

Preferred to collecting CO₂ from fossil power stations

R&D Blocked by NO NUKES

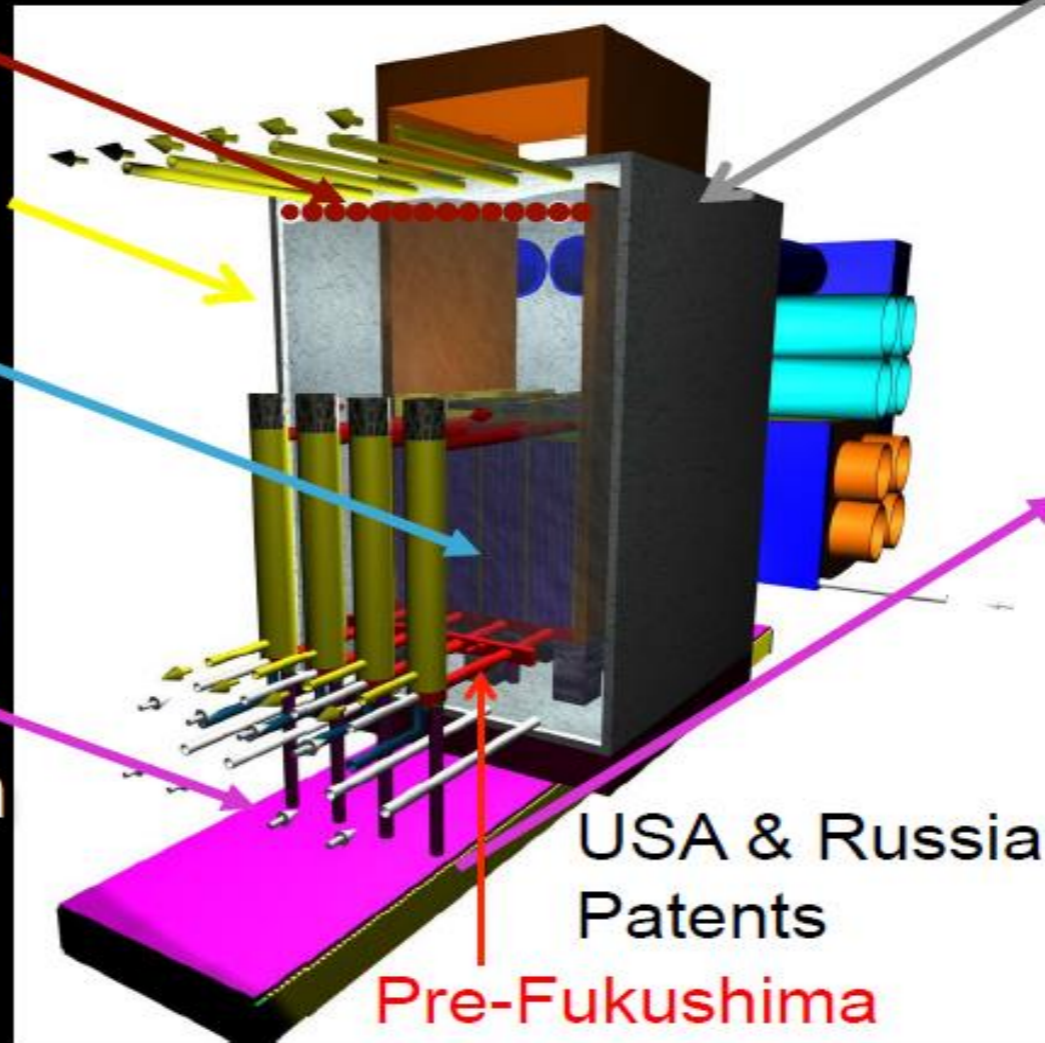
No money and No political will to develop and use safer nuclear power

Molten Salt Reactor

ORNL 1960s, HX 2009, Transatomic 2010, Flibe 2011, CAS 2011

Reactor core built from graphitic materials

Automatic descent of control elements if LOCA, no Chernobyl
If over-heated, fuel salt expands out of reaction zone. If T still rises, frozen plug melts; fuel salt drains into dump tank, which is air-cooled to remove decay heat (cannot lose air).



2 metal walls
He & HF to remove Xe & I
For other FPs, vacuum distill
Spill: CsF, SrF₂ in salt that freezes in < 10 s
Solid salt inert, low vapor P : no radioactive gases, no fires, no explosions.

Thick steel dome, no Chernobyl, no Fukushima, no jet crashes.
Walk-away safe w/o operator intervention. 100 MWt, extra safe.

Safe Reactor Design and Construction and Experience too expensive, even for those countries still interested. Safe reactors will not be ready when needed.

The Amazon rain forest is disappearing at a speed of approximately 4000 football fields per hour!



- Wood for construction and for making fires.
- Agriculture (soy) for farms.
- Land for poor farmers who have no other place to live.
- Grazing land for cattle.
- Road construction
- Oil production
- Mining



Droughts are changing the Amazon rainforest

Droughts are increasingly frequent in the Amazon rainforest, with three "100 years" droughts in the space of only 10 years. Trees have openings at the bottom of their leaves called stomata, which open and close to regulate the amount of carbon dioxide a tree absorbs and the amount of water vapor it releases. When there is more carbon dioxide in the atmosphere, stomata do not need to open so widely and, therefore, release less water vapor, and fewer clouds will form over the forest.



The Amazon rainforest suffered severe droughts that killed large amounts of vegetation in the most affected areas. Also, unfortunately, an increase in global temperature can kill many more trees.



1950



1985



2000



2005



2010



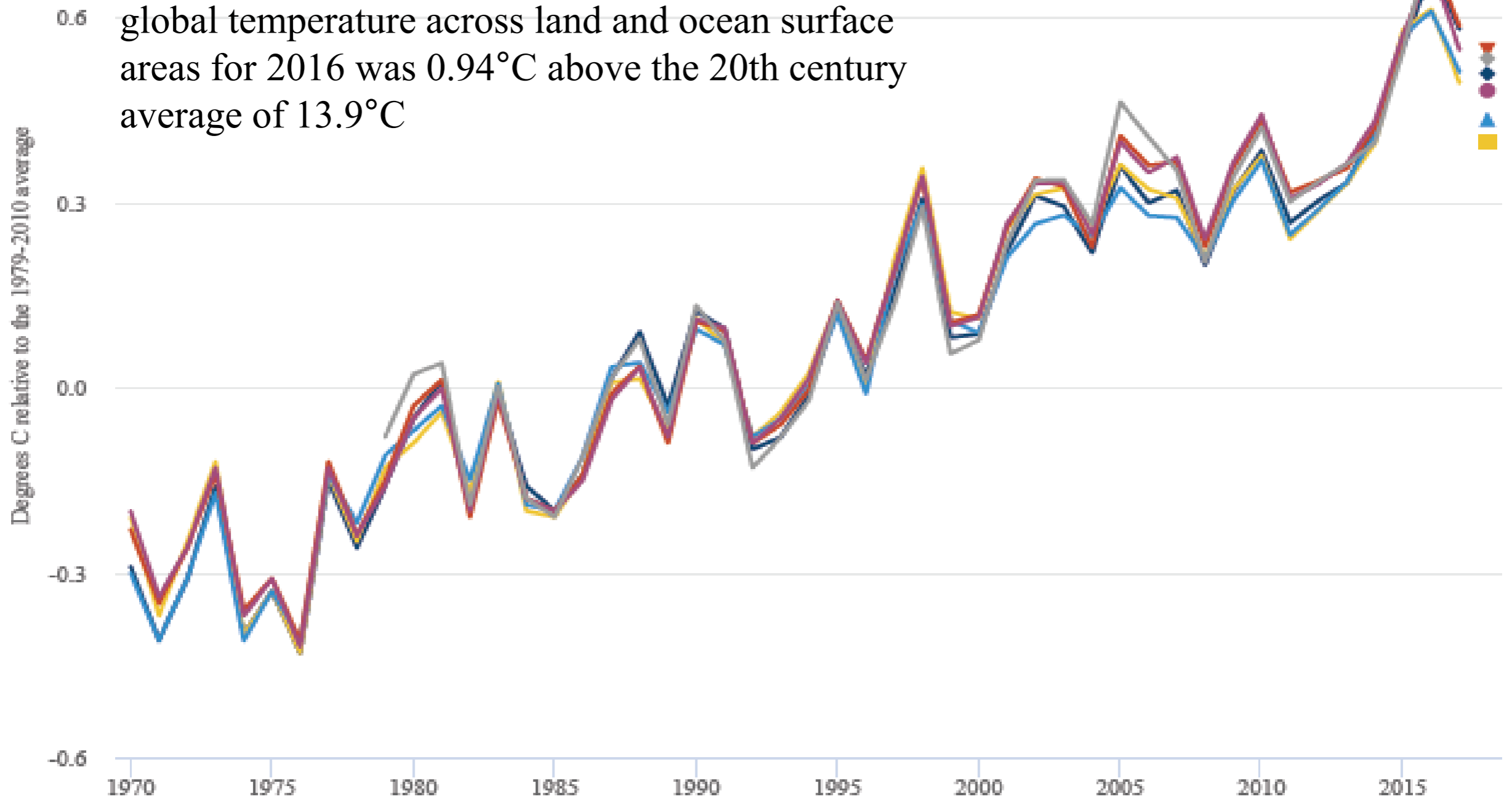
2020



Global surface temperature records, 1970-2017, and 2018 to-date

— NASA GISTemp — Hadley/UEA HadCRUT4 — NOAA GlobeTemp — Berkeley Earth
— Cowtan and Way — Copernicus/ECMWF

NOAA's global analysis: 2016 became the warmest year in NOAA's 137-year series. The average global temperature across land and ocean surface areas for 2016 was 0.94°C above the 20th century average of 13.9°C

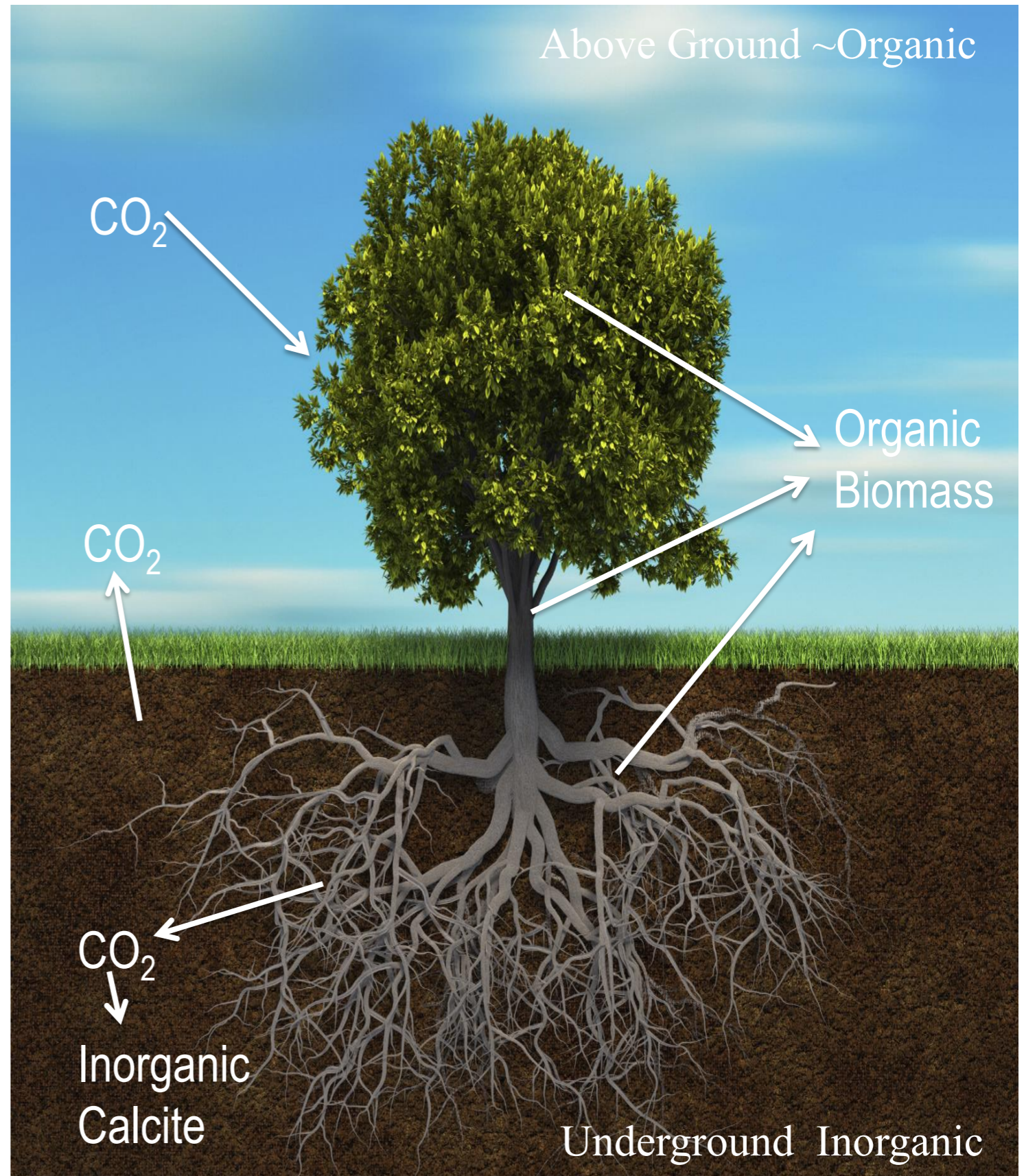




2015 Paris Climate Agreement

- The 2015 Paris Agreement deals with GHG emissions mitigation, adaptation and finance starting in the year 2020, to help stop the gradual warming of the Earth. 2018 Santiago, Chile meeting Dec. 2018.
- Main aim is to limit the increase in the global average temperature to well below 2°C above pre-industrial levels, by reducing GHG emissions.
- But this target now requires "negative emissions", i.e., extraction of CO₂ from the atmosphere.
- By July 2018, 194 states and the European Union have signed the Agreement.
- **But a study published 2017 in Nature found that all major industrialized nations are failing to meet the pledges they made in the Paris Agreement.** And these pledges are anyhow inadequate. Emissions keep rising and impacts are being felt all around the world. USA, India and China were responsible for 85% of the global rise in energy-related carbon dioxide emissions over the past year.
- USA for example delivered an official notice to the UN that it intends to withdraw from the Paris Agreement 4 Nov. 2019, the earliest date that it is legally eligible to withdraw.

- Photosynthesis - Leaves inhale CO_2 . Roots in semi-arid regions are deep, and also exhale CO_2 into USZ at high partial pressure
- $\text{CO}_2 + \text{H}_2\text{O} \rightarrow$ dissolved HCO_3^- bicarbonate, combines with soil Ca^{+2} , precipitates CaCO_3 calcite.
- Trees in semi-arid regions sequester atmospheric CO_2 long term underground as stable calcite, low rainfall – calcite does not dissolve.
- These trees also sequester atmospheric CO_2 short term, as organic biomass.
- But burning, decomposition of biomass reinject CO_2 into the atmosphere.



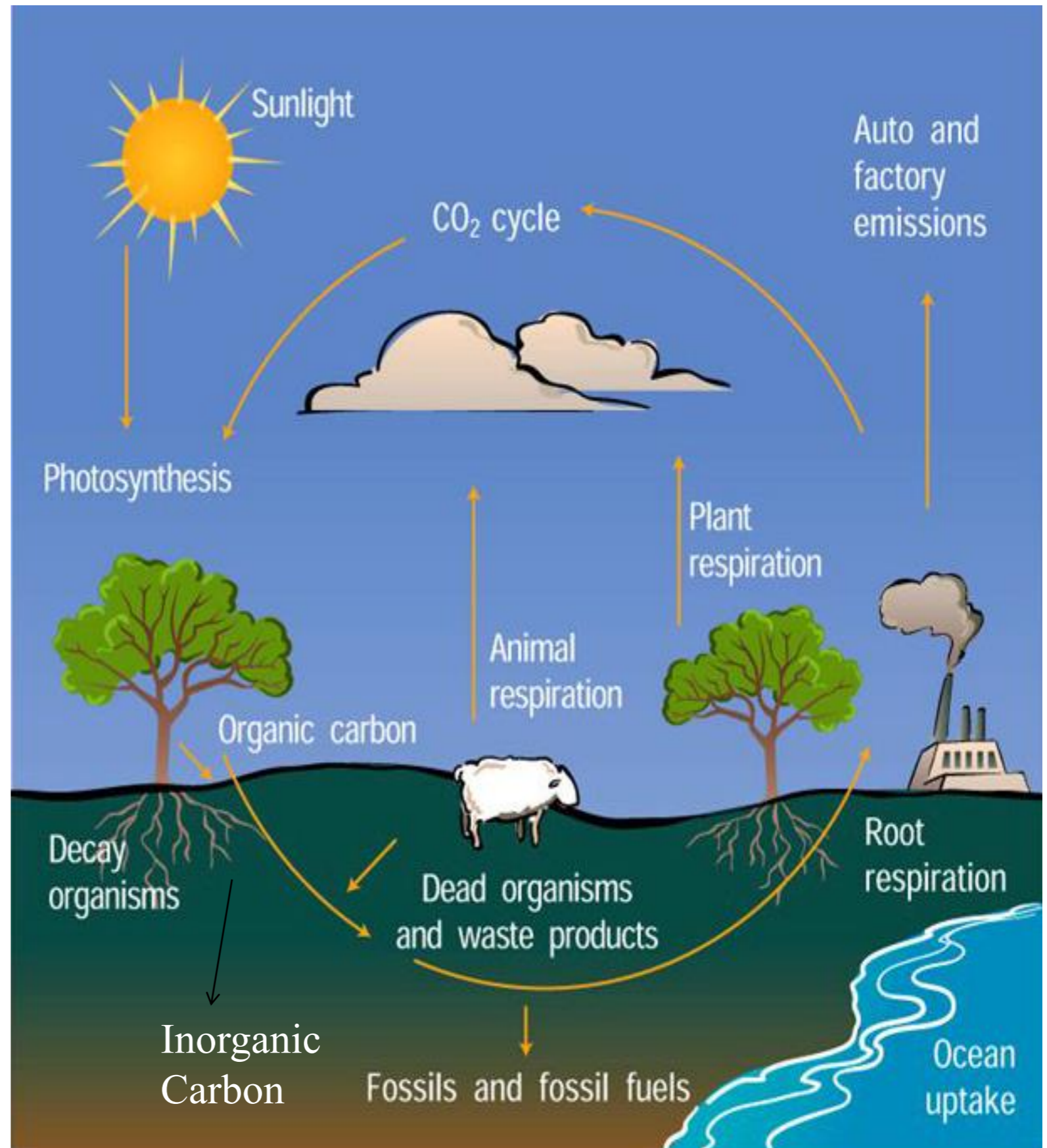


Tree Respiration

- **Why is CO₂ exhaled by the roots?**
- During the day, the tree inhales air (CO₂, O₂, etc.) through the stomata. and uses CO₂ via photosynthesis to produce O₂ and C₆H₁₂O₆ glucose. Some inhaled or produced O₂ is used to oxidize glucose, to produce energy for the tree's biological functions, via the reaction
$$\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$$
- Not all the glucose is oxidized, but is rather used for other biological purposes. The tree's metabolism is such that the oxidation of glucose, and tree growth, work at a higher rate at night, when the tree is not occupied with photosynthesis. So the roots exhale CO₂ produced by oxidation of the glucose. Besides CO₂ sequestration, the trees supply needed O₂.

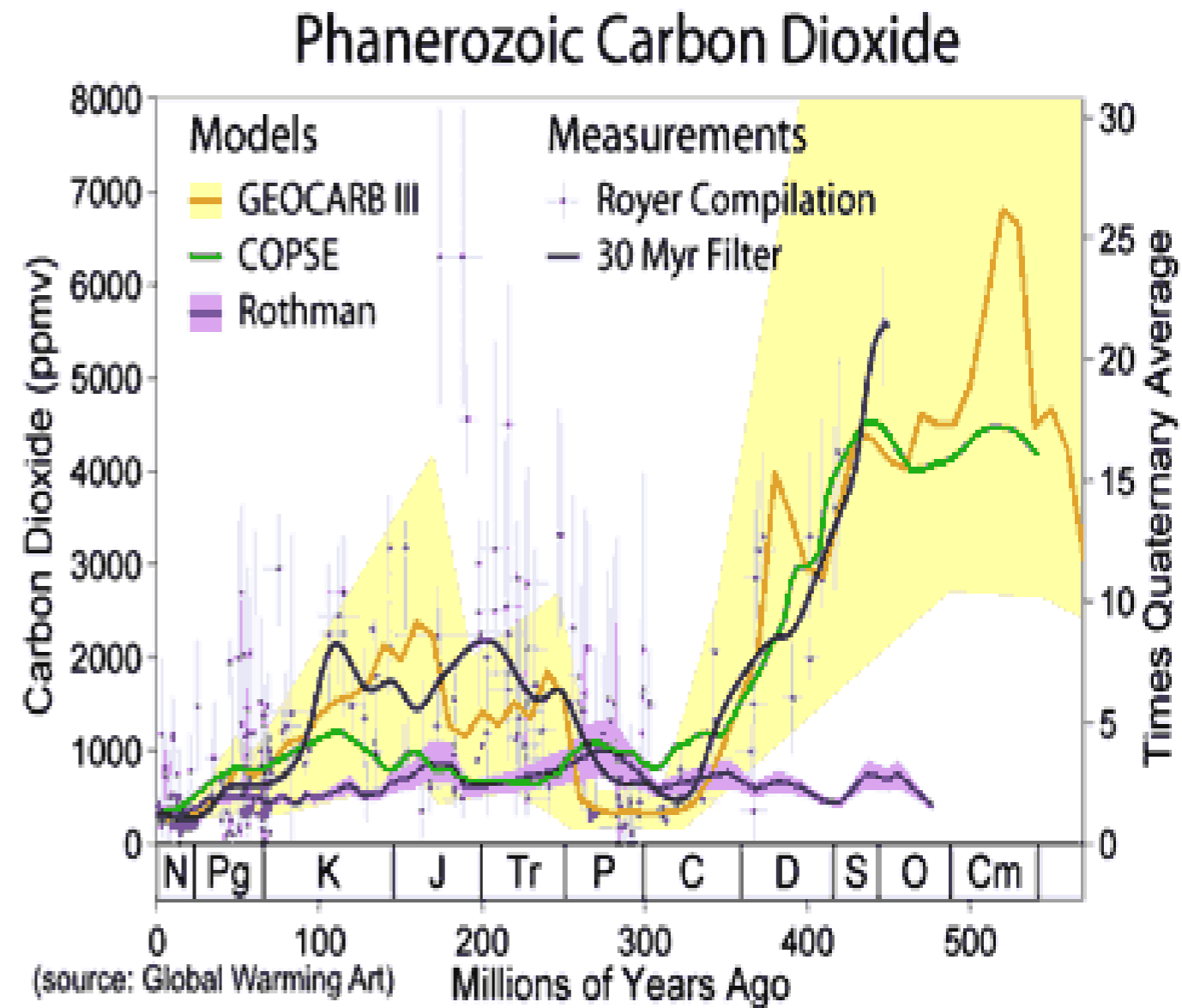
The Carbon Cycle

- Burning fossil fuels and decomposition of dead plants and animals inject CO_2 into the atmosphere.
- Organic carbon production via forestation provides useful short term organic carbon sequestration.
- Semi-arid forests remove atmospheric carbon dioxide, fixing it underground long term as inorganic carbonate salts.



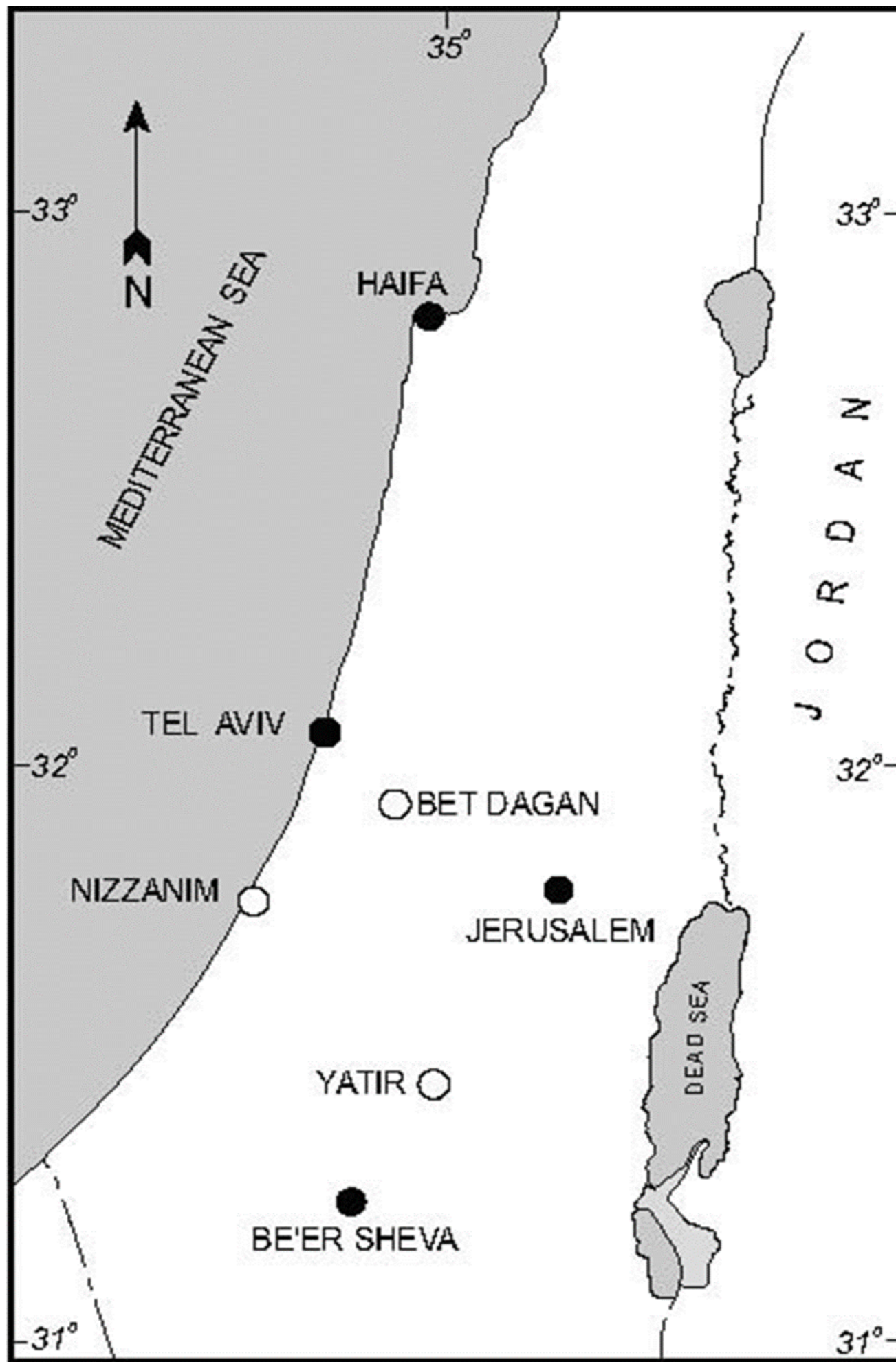
Is there evidence that forestation can efficiently sequester carbon dioxide?

Yes! Atmospheric CO₂ was ~15 times present concentrations during the middle Devonian by volcanism. During the following Carboniferous period, biomass from great forests was preserved within carboniferous coal and oil deposits. This carbon is being returned to the atmosphere as fossil fuels are burned. The transfer of CO₂ to plant biomass tapered off with the onset of the late carboniferous Permian Karoo Ice Age. The figure shows the variability of atmospheric CO₂ throughout the Phanerozoic. Notwithstanding the great increase since the Industrial Revolution, atmospheric CO₂ levels at present are relatively low on a geological time scale.

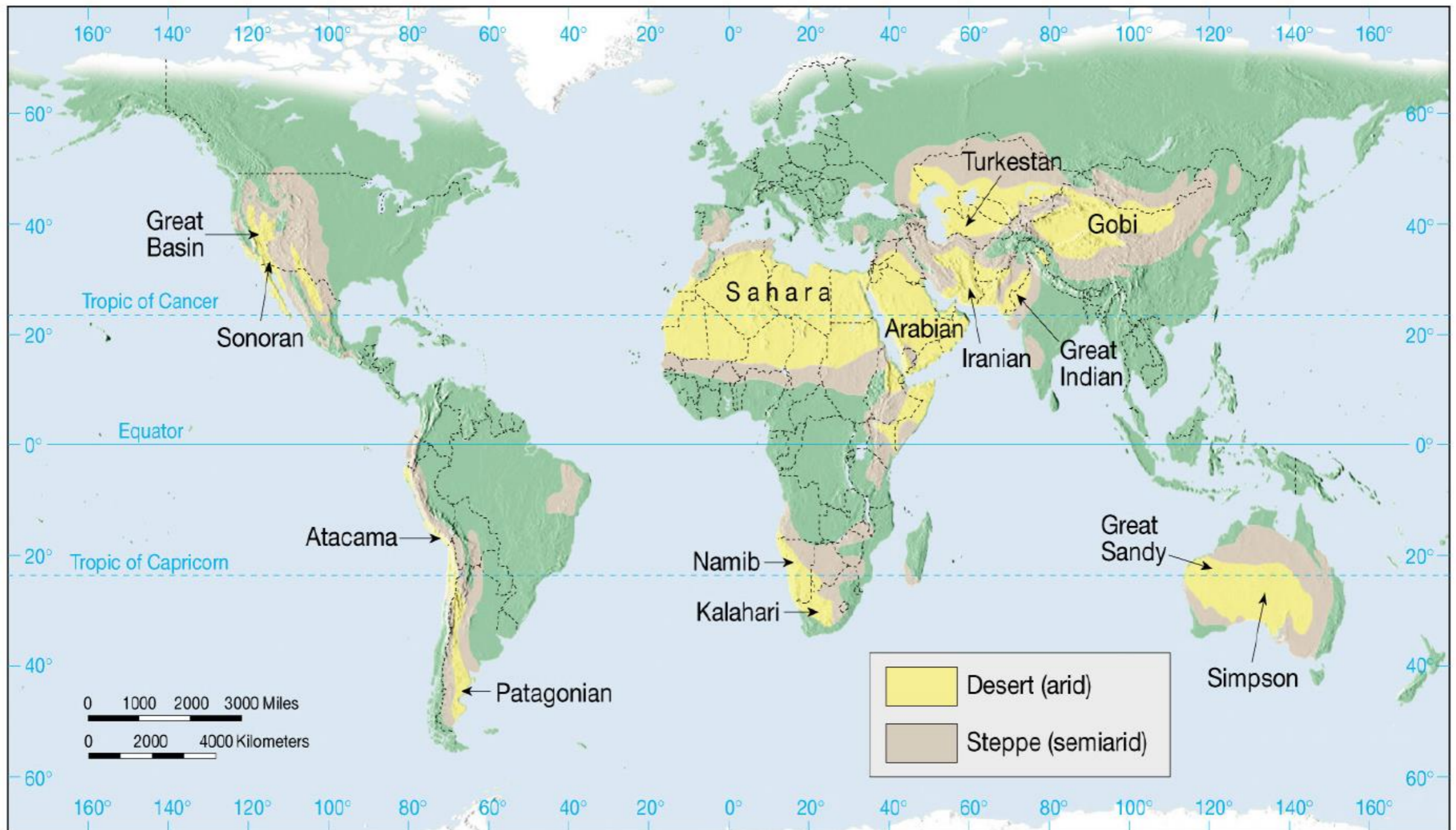




Yatir forest, semi-arid ~ world's driest - 28 cm per year - largest forest in Israel, 28 km² (6920 acres or 2,800 hectares), C3 Type. Sediment is loess (windblown, fine quartz sand, clays, fine grain limestone). Semi-arid is 18% of global land surface.



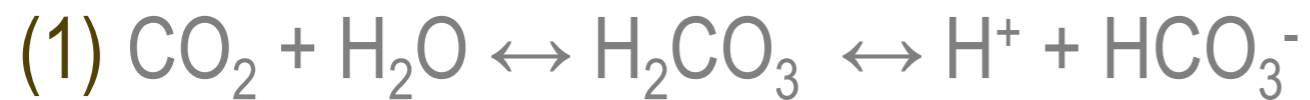
Global distribution of semi-arid (Steppe, 17.7%) regions





Chemical Equations for Carbonate Precipitation

Soil gas dissolves into the soil moisture, forms carbonic acid.



Carbonic acid dissolves pre-existing soil calcite and releases calcium ions.



Released Ca^{+2} combines with the DIC, and precipitates calcite.



Eq. 3 (idealized) implies no net sequestration. Our isotopic data show net sequestration because: (a) most CO_2 from Eq. 3 does not return to atmosphere, maybe only 20%, roots are deep, most remains in soil gas (b) many other cation sources contribute to Eq. 3, desorption of Ca^{+2} from clay, gypsum, sea spray, etc.



Yatir forest soil cores

- Porosity (~53%), Humidity (~12%), Dry sediment density (1.24 g cm^{-3}) → A Liter of sediment has 0.47 L solid + 0.12 L water + 0.41 L gas.
- Measure CO_2 precipitation rate (into CaCO_3) per cm per Liter of sediment (L_s) from decrease per cm of DIC ($\sim 2.0 \text{ mg cm}^{-1} L_s^{-1}$ at Yatir).
- Measure flow rate of water (11 cm yr^{-1} at Yatir) using tritium in water (HTO) as tracer.
- CO_2 precipitation rate (within calcite CaCO_3) at Yatir is $\sim 22 \text{ mg yr}^{-1} L_s^{-1}$ of CO_2 ($11 \text{ cm yr}^{-1} \times 2.0 \text{ mg cm}^{-1} L_s^{-1}$). Calcite is derived from CO_2 exhaled by tree roots. The Carbon isotope fractionation $\delta^{13}\text{C}$ in the DIC is due Yatir's C3 type plants that exhale CO_2 with $\delta^{13}\text{C} \sim -22\text{‰}$.

Dissolved Inorganic Carbon (DIC) vs. Depth

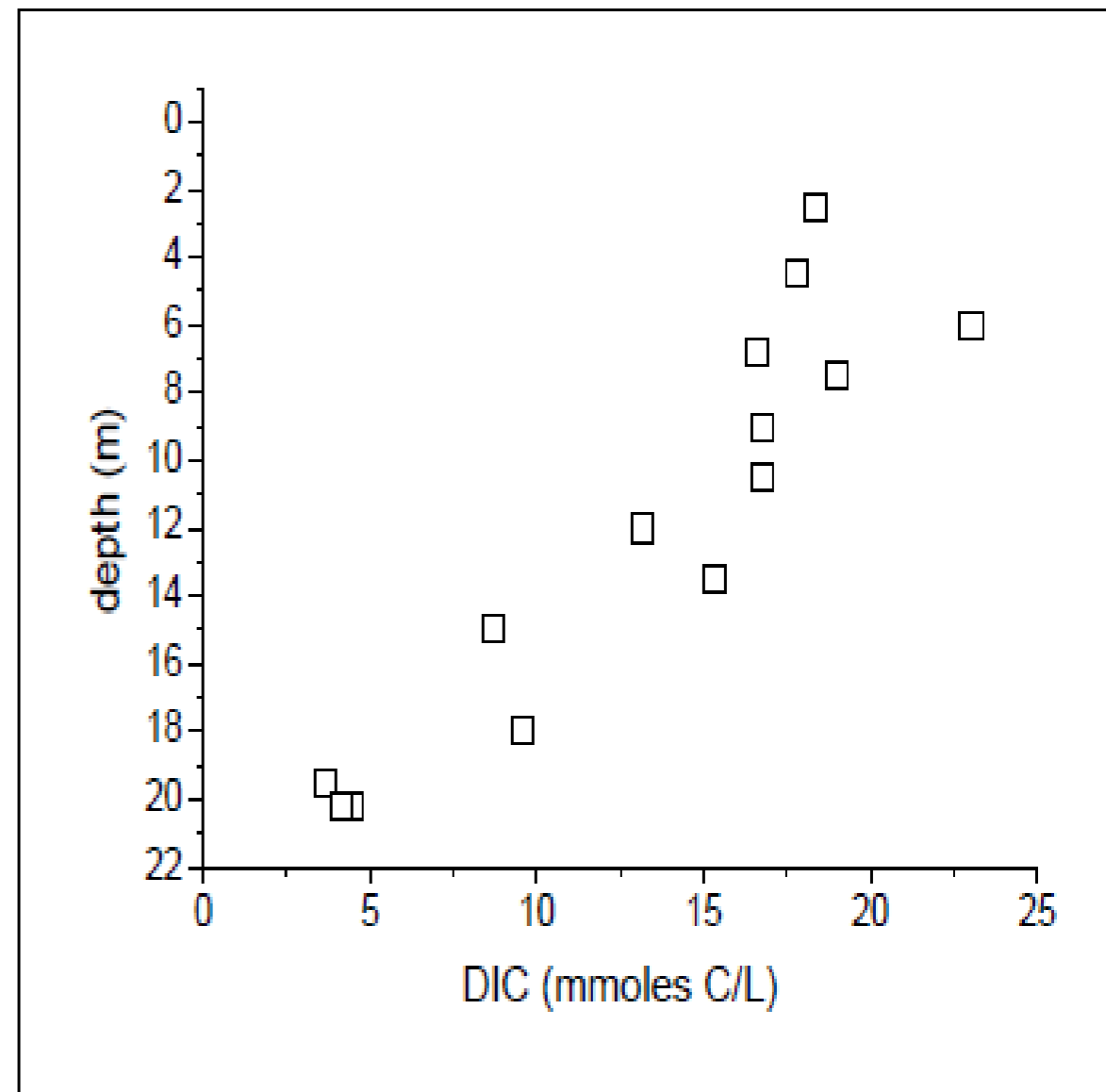
Measure DIC (as CO_2) from CO_2 mass & liters of water (L_w) extracted from USZ wet sediment by vacuum distillation ($\sim 250 \text{ mg CO}_2 L_w^{-1}$ at Yatir)

Yatir Plants

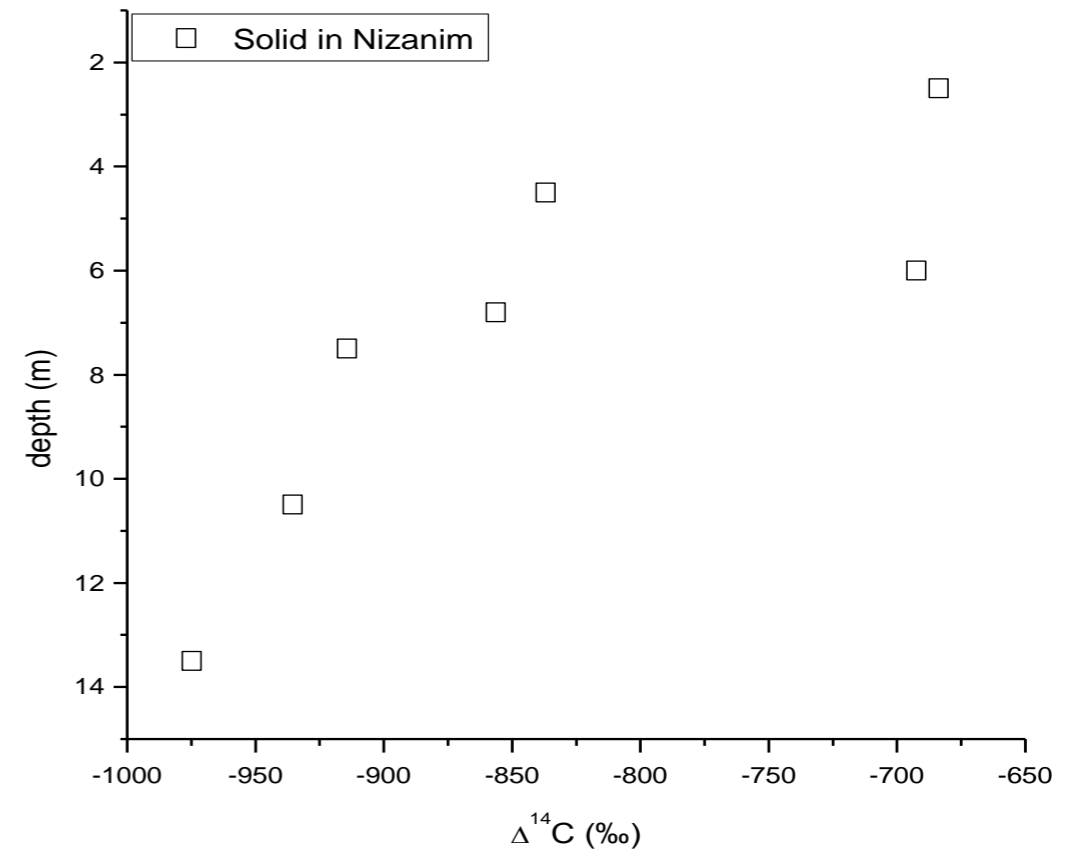
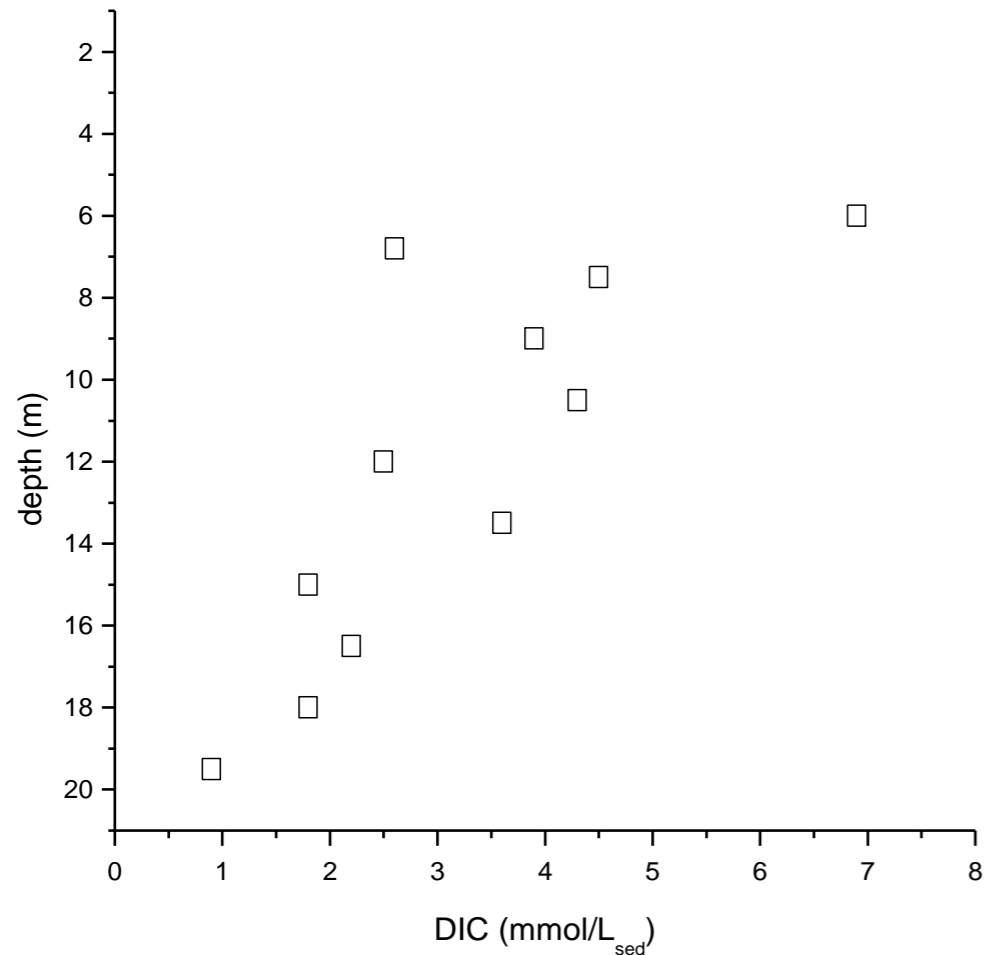
Depth (cm)	DIC (mmol C L^{-1})
30-60	4.5
60-90	3.4

CO_2 from roots combines with soil water of unsaturated zone (USZ) to form DIC, comprised mainly of HCO_3^- bicarbonate. DIC concentration decreases with depth, as water flows down, due to CaCO_3 precipitation into USZ. Seen easily for plants with shallow roots, not tree roots.

Nizzanim Plants



Nizzanim DIC and ^{14}C

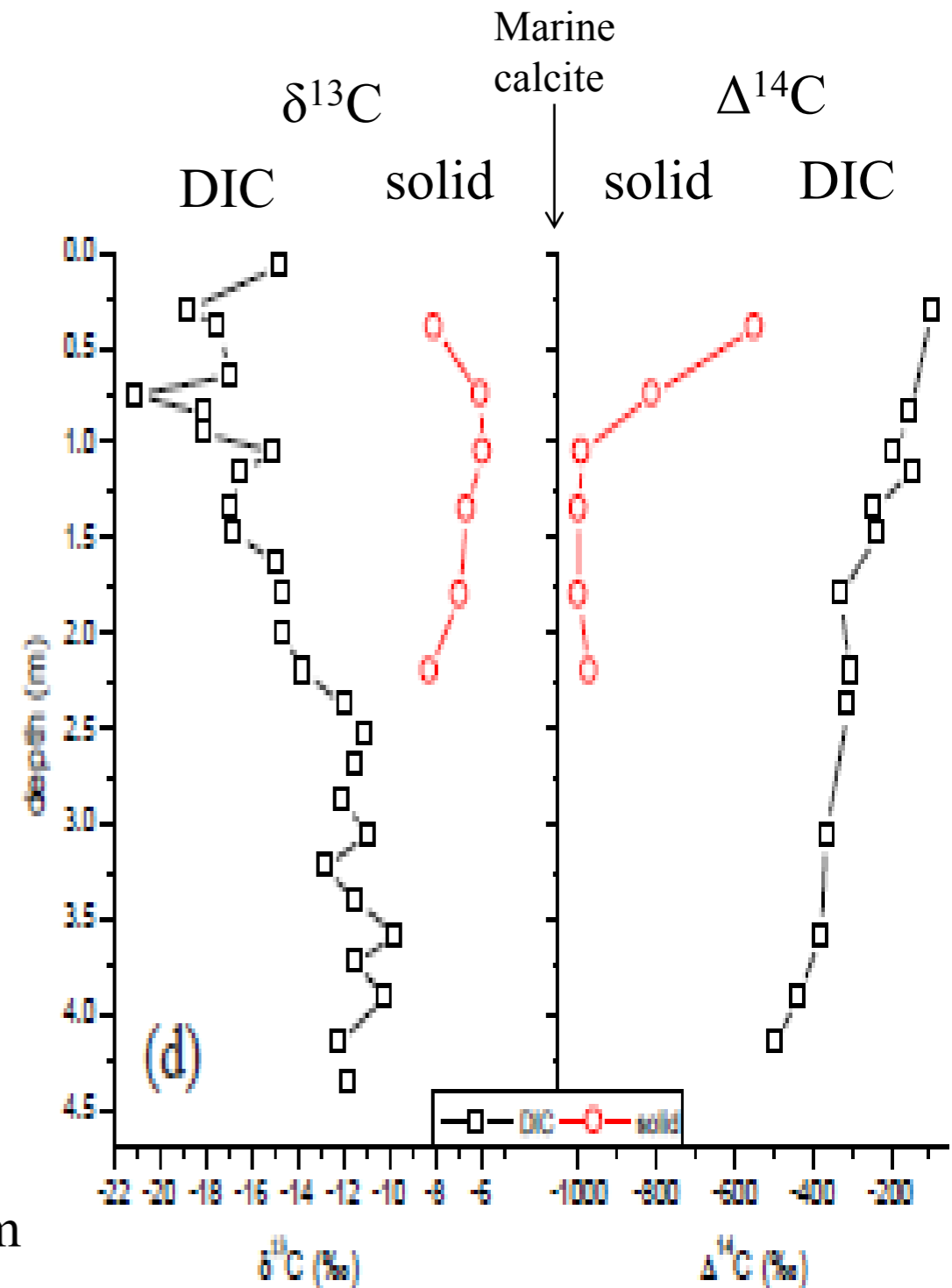


Left: DIC decreases with depth, due to removal of carbon from solution and exchange with sediment.

Right: ^{14}C activity in the sediment is from atmospheric CO_2 injected into the DIC by the overlying plants, as the host calcite is too old to contribute radiocarbon. Once $^{14}\text{CO}_2$ enters the DIC, a continuous carbon isotopic exchange and precipitation occurs between the DIC and the solid carbonate. The ^{14}C activity in the solid decreases as a function of depth, correlated with the decreasing DIC concentration,

Isotope exchange in Yatir (and Nizzanim) forests

- ^{13}C and ^{14}C are tracers for precipitation reaction
- Soil contains up to 20% relict marine calcite.
- DIC concentration decreases with depth.
- Relict calcite: $\Delta^{14}\text{C} = -1000$ (zero ^{14}C).
- Relict calcite: $\delta^{13}\text{C}=0$ (same ^{13}C as standard)
- Continuous exchange between DIC and marine calcite in USZ as rainwater flows down.
- $\delta^{13}\text{C}$ and $\Delta^{14}\text{C}$ values in solid (nodules) vary due to exchange between DIC and relict calcite; & precipitation onto sediment nodule surface.



^{13}C and ^{14}C in the **solid** and liquid phases of the sediment. The solid phase shows that modern ^{14}C had precipitated from the liquid on the **solid** phase



Isotopic evidence for CO₂ incorporation and sequestration into pedogenic carbonate

- $\delta^{13}\text{C}$ and $\Delta^{14}\text{C}$ trace the incorporation of atmospheric CO₂ from DIC into pedogenic carbonate, as DIC descends down profile interacting with relict marine limestone
- Depleted $\delta^{13}\text{C}$ values in DIC from atmospheric CO₂ - exchanges (continuous dissolution and precipitation) with $\delta^{13}\text{C}$ -enriched relict marine limestone to form pedogenic calcite (**solid**) of intermediate $\delta^{13}\text{C}$ values
- Atmosphere is the sole source of ¹⁴C, introduced into the USZ as ¹⁴CO₂ via roots and decay
- Pedogenic **solid** in the USZ incorporates ¹⁴C, as DIC exchanges with relict limestone
- Therefore: atmospheric CO₂ tagged by radiocarbon is incorporated into pedogenic carbonate
- Not previously observed. AMS allows C14 measurement with 2 ml.



Global Extrapolation of Inorganic Carbon Precipitation in Forest Soil Sediment

- **CO₂ precipitation rate (in calcite CaCO₃) at Yatir is ~22 mg yr⁻¹ L_s⁻¹ of CO₂ (11 cm yr⁻¹ x 2.0 mg cm⁻¹L_s⁻¹). Calcite is derived from CO₂ exhaled by tree roots. The Carbon isotope fractionation δ¹³C in the DIC is due Yatir's C3 type plants that exhale CO₂ with δ¹³C ~ -22‰.**
- Consider sediment volume 1 km² & 6 m depth (6x10⁹ L), trees above, roots below. **Precipitation ~130 tons of CO₂ per year (22. x10⁻³ x 6x10⁹ gm).**
- **Extrapolating semi-arid sequestration rate globally yields rate of ~3.0 billion tons inorganic CO₂/yr (~130 tons km⁻² x 24 x 10⁶ km² ~ 3.0 Pg), ~15% of the rate by which atmospheric CO₂ is currently increasing.**
- In addition, the sequestration of organic carbon following global semi-arid forestation has been estimated by other investigators as **~2 billion tons of organic CO₂ each year**. All others only consider organic.
- **A particular global land management policy is tentatively suggested - widespread tree planting in semi-arid regions.**



Forestry Carbon Reduction Credits

- 6.2 billion tons emission allowances and offsets were traded globally in 2015; total value was €48.4 billion. 3.0 billion tons of CO₂ via inorganic sequestration are potentially worth ~\$30-\$90 billion in CRCs. One **carbon credit** is equal to one ton of **carbon** dioxide.
- Forests may generate carbon reduction credits (CRC) via afforestation and/or reforestation (AR), improved forest management, conservation and protection against deforestation. However, forestry sequestration activities recognized under the Clean Development Mechanism of the Kyoto Protocol (CDM) are restricted to AR. The AR CDM is a mechanism to credit carbon sequestration by forests.
- The difference between a **baseline scenario** and the **project scenario** indicates the carbon sequestration achieved through the AR activity and the resulting amount of tradable carbon credits. The baseline scenario describes the development of carbon stocks without the CDM project, through the existing or historical changes in carbon stocks in the vegetation. The project scenario estimates the effect of AR on the annual increase in carbon stocks.



Advantages of Semi-Arid Forest Sequestration

Advantages of inorganic carbon sequestration into USZ in semi-arid regions via *Forestation*

- **Trees can be harvested** for useful products (lumber, charcoal, off-sets, etc.); while **inorganic carbon deposition remains underground**. Harvested biomass will be an important future renewable energy component, as fossil fuel supplies decrease.
- Extensive thickness of USZ (deep roots) compared to temperate regions – therefore greater active volume.
- **Low rainfall precludes dissolving calcite**, as in temperate regions.
Yatir forest receives only ~28 cm/yr rainfall. World's **driest and healthy forest**.
- Semi-arid afforestation does not reduce food supply by decreasing productive temperate-region agricultural land
- Provide steady employment in economically depressed areas instead of marginal herding & agriculture



Suggested International Collaborations

- Synergy- Tel Aviv geoscientists with inorganic carbon sequestration expertise and collaborators with forestation and organic sequestration and legal expertise can help limit global warming.
- Research aims include measurements in Israel and other countries to help achieve an improved estimate of potential global sequestration rates.
- May also lead to establishing business partnership, based on filing world wide patents, with an appropriate percentage of patent rights assigned to capital investors.
- Business objective is to develop, create and commercialize optimized method of evaluating a Forest's Inorganic Carbon Sequestration Rate.



Sequestration Summary

- **Atmospheric derived CO₂ sequestered long-term as inorganic calcite in USZ of semi-arid forests - much longer-term carbon reduction and storage compared to organic carbon biomass.**
- **Global extrapolation yields CO₂ annual semi-arid sequestration rate of ~3.0 billion tons inorganic CO₂/yr, plus ~2,0 billion tons organic CO₂/yr. The total carbon sequestration rate after global semi-arid forestation then potentially represents a respectable ~25% of rate of atmospheric CO₂ increase.**
- **We developed the technology to measure and certify inorganic carbon sequestration rate of semi-arid forests.**
- **Forests may thereby generate valuable annual carbon reduction credits. Economically sustainable & profitable. The forests also provide needed O₂**

Semi-arid forest related biomass will be **an important future renewable energy component**, as fossil fuel supplies decrease (lumber, charcoal, biochar, etc.).

Further measurements now in progress at Israel's Ilanot Forest – Looking for collaborators in other semi-arid regions. **Need more data** to better characterize the potential global sequestration rate.

A photograph of a forest with tall, thin trees. The trees are mostly bare, with some sparse green foliage visible in the upper right. The scene is lit with a warm, golden light, creating a soft glow on the tree trunks and branches. The background is dark, making the trees stand out.

Thank You !